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Half-sandwich complexes of Ir(III), Rh(III) and Ru(II) with the MaxPhos ligand: metal centred chirality and cyclometallation†

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The reaction of the acetylacetonates $[(\eta^5\text{-C}_5\text{Me}_5)\text{M}(\text{acac})\text{Cl}]$ with $(S_P)\text{-[HMaxPhos][BF}_4]$ afforded cationic complexes with the formula $(S_M,R_P)\text{-}[(\eta^5\text{-C}_5\text{Me}_5)\text{MCl}(\text{MaxPhos})][\text{BF}_4]$ ($M = \text{Rh}$ (**1**), Ir (**2**)). The reaction of $(S_P)\text{-MaxPhos}$ with $[\text{RuCl}(\mu\text{-Cl})(\eta^6\text{-}p\text{-MeC}_6\text{H}_4\text{Pr})_2]$ and NH_4X afforded $(S_{R_u},R_P)\text{-}[(\eta^6\text{-}p\text{-MeC}_6\text{H}_4\text{Pr})\text{RuCl}(\text{MaxPhos})][\text{X}]$ ($X = \text{BF}_4$ (**3**), PF_6 (**3'**)). The complexes have been completely characterized by analytical and spectroscopic means, including the determination of the crystal structures of **1**, **2** and **3'**. Treatment of the iridium complex **2** with AgBF_4 , at 253 K, resulted in the intramolecular cyclometallation of one of the *tert*-butyl substituents of the MaxPhos diphosphane ligand, affording a mixture of isomers of $(S_{\text{Ir}},R_{P1},S_{P2})$ and $(R_{\text{Ir}},R_{P1},R_{P2})\text{-}[(\eta^5\text{-C}_5\text{Me}_5)\text{Ir}(\text{MaxPhos})][\text{BF}_4]$ (**4a** and **4b**). However, rhodium complex **1** and ruthenium complex **3** reacted with AgBF_4 forming the expected unsaturated intermediates $(\eta^{\text{ring}}\text{-ring})\text{M}(\text{MaxPhos})$ which were trapped by MeCN, affording the cationic adducts $(S_M,R_P)\text{-}[(\eta^{\text{ring}}\text{-ring})\text{M}(\text{MaxPhos})(\text{MeCN})][\text{BF}_4]_2$ ($M = \text{Rh}$ (**5**), Ru (**6**)). Only one epimer at the metal was isolated in high yield for the complexes **1**, **2**, **3**, **3'**, **5** and **6** and the metallation of **2** to give **4** occurs with high diastereoselectivity.

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Introduction

Since the origins of asymmetric homogeneous catalysis with transition metals, chiral phosphanes and their derivatives have been the preferred type of ligand.^{1–3} The synthesis of new phosphane ligands is still an extremely active area, fostered by a continuous need for new ligands that lead to faster and more selective processes, operating under milder conditions. This research is leading to new families of chiral phosphanes, exploiting all possible sources of diversity including stereogenic carbon and phosphorus atoms, axes and planes.^{4,5} A par-

ticularly appealing family is that of *P*-stereogenic phosphanes, a ligand type that, despite being linked to the origins of asymmetric catalysis, was neglected for a long time⁶ and is now experiencing an intense renaissance.^{7–10}

A few years ago, some of us reported on the MaxPhos ligand (see Scheme 1), which is a rare example of a *P*-stereogenic, C_1 -symmetric ligand with a short bridge between the phosphorus atoms.^{11,12} In addition, thanks to a tautomeric equilibrium involving the P atoms and the NH bridge, it was obtained as its bench-stable crystalline tetrafluoroborate salt, in contrast to many of the new generation highly basic and air-sensitive *P*-stereogenic ligands with alkyl substituents.¹³

The coordination chemistry of this ligand was initially explored with Rh(i) diene complexes,¹¹ which were found to be

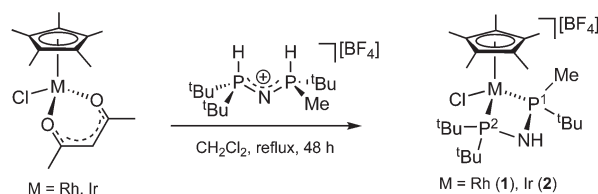
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† Electronic supplementary information (ESI) available: Selected NMR, MS, IR and CD spectra of the compounds. X-ray crystallographic information files containing full details of the structural analysis of complexes **1**, **2** and **3'**. CCDC 1573023–1573025. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c7dt03327c



Scheme 1 Reaction of $(S_P)\text{-[HMaxPhos][BF}_4]$ with Rh- and Ir-acac precursors.